Millimeter Wave Rotational Spectrum and Centrifugal Distortion of Thioketene, H₂C=C=S

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a-type rotational transitions of molecules in the vibrational ground state of thicketene, $H_2C=C=S$, have been measured in the millimeter wavelength region. The measurements yielded improved rotational constants:

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A = 286 655(82) \text{ MHz},

B = 5 659.47596(72) \text{ MHz},

C = 5 544.51269(72) \text{ MHz}.
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A detailed centrifugal distortion analysis by means of Watson's S-reduced Hamiltonian led to the determination of four quartic, two sextic and two higher order distortion constants:

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D_J = 1.08569(4) \text{ kHz}, \qquad H_{JK} = 0.716(20) \text{ Hz}, \ D_{JK} = 168.269(77) \text{ kHz}, \qquad H_{KJ} = -408.7(73) \text{ Hz}, \ d_1 = -25.46(68) \text{ Hz}, \qquad L_{KJ} = 0.65(24) \text{ Hz}, \ d_2 = -5.21(35) \text{ Hz}, \qquad S_{KJ} = -0.0533(24) \text{ Hz}.
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Effective rotational and centrifugal distortion constants using planarity conditions were calculated. The electric dipole moment of thicketene was determined to be $\mu = 1.01(3)$ D.

I. Introduction

Intensive studies have been carried out on the rotational spectra of various isotopic species of ketene, $H_2C=C=0$, since 1950 [1-6]. The centrifugal distortion analysis of ketene [3, 4], together with information from additional infrared data [3, 5], enabled Mallinson and Nemes [6] to reevaluate the molecular force field for ketene. Recently, microwave investigations on the higher homologues thioketene, $H_2C=C=S$ [7, 8], and selenoketene, $H_2C=C=S$ [9, 10], have been reported.

The first interstellar microwave detection of interstellar ketene was reported in 1977 by Turner [11]. The detection in the source Sgr B2 was based on three transitions: $5_{14}-4_{13}$, $5_{15}-4_{14}$ and $4_{13}-3_{12}$ at the frequencies 101 981.39, 100 094.51 and 81 586.19 MHz, respectively. Further millimeter wave transitions were tentatively detected and assigned. Recently, Thaddeus and coworkers have identified isothiocyanic acid, HNCS [12], and methyl mercaptan, CH₃SH [13], in Sgr B2 from observations of a number of millimeter wave

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rotational transitions. The abundances of both of these sulfur-containing molecules, relative to their oxygen analogues which had previously been detected in interstellar dust clouds, imply a sulfur/oxygen ratio not very different from the terrestrial value. Therefore, the detection of the five-atomic molecule $H_2C=C=S$ in interstellar dust clouds might become possible with more sensitive instrumentation. The predicted rotational transition frequencies based on the measured millimeter wave spectrum, together with calculated standard deviations, provide the basis for a possible search for interstellar thioketene.

Moreover, thicketene has raised some interest due to the fact that it is a more stable isomer of thiirene, H-C=C-H, according to SCF-calculations [14,

15]. These calculations also reveal the fact that ethynylmercaptane, $H-C\equiv C-SH$, should have the same stability as thioketene. As a matter of fact, Krantz and Laureni [16] detected both ethynylmercaptane and thioketene in the photolysis of 1,2,3-thiadiazole. However, no ethynylmercaptane was found in the products of pyrolysis reactions [14]. This was confirmed by our microwave experiments.

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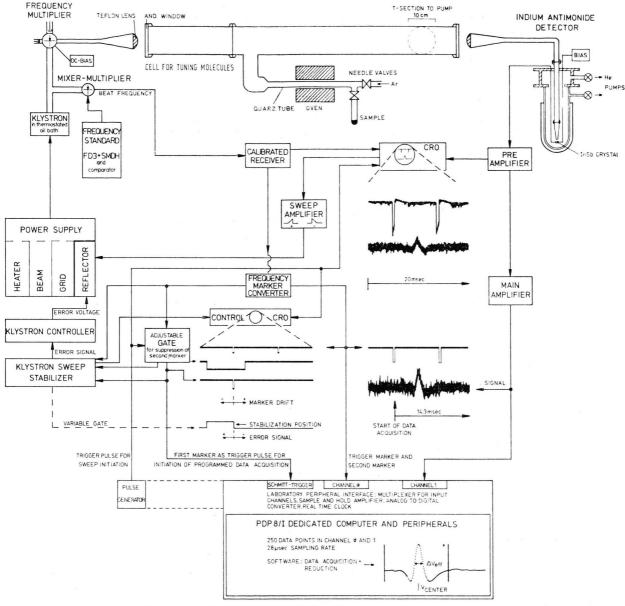


Fig. 1. Block diagram of the improved computer controlled millimeter wave spectrometer as discussed in [21].

Several methods of preparing thioketene are known [17]. All reports confirm its instability at room temperature, so that only pyrolysis experiments in flow-systems can be taken into consideration for spectroscopic experiments. In such an experiment Georgiou, Kroto and Landsberg measured four rotational transitions and could determine the rotational constants B and C and the centrifugal distortion constant D_{JK} of thioketene [7].

In this paper a-type R-branch ground state transitions are reported which have been studied in the millimeter-wave region from 60 up to 230 GHz. 125 transitions have been measured and fitted with the S-reduced Hamiltonian according to Watson [18].

II. Experimental Procedures

Thioketene was produced in two different ways (Scheme I).

As in the report on the earlier microwave measurements of thioketene [8], hexamethyltrithiane (I) was pyrolysed at 1300 K to give thioketene, which was directly pumped through the 2 m glass cell of the millimeter wave spectrometer as shown in Figure 1. Infrared analysis of the gas mixture after pyrolysis showed that a large fraction of the products consisted of ethylene, propylene and allene, thus indicating that only a rather poor yield of thioketene is obtainable by this method. A recently reported procedure [14] of preparing thioketene by pyrolysing dithioacetic acid (Scheme I) was found to yield a considerably higher intensity of the absorption lines, thus indicating higher column densities of H₂CCS.

Attempts to isolate thioketene have failed so far. It could be trapped (spectrum disappears) at temperatures below about 150 K, but polymerised upon warming. Even in the gas phase, at sample pressures of 40 µbar, decay took place with a halflife of about five minutes, independent of the way of preparation. The use of argon as carrier gas in order to entrain the thioketene did not improve the stability.

Microwave measurements were performed near 22 GHz in order to determine the dipole moment using a Hewlett-Packard spectrometer, 8460 MRR. The millimeter-wave measurements were carried out with a spectrometer described by Winnewisser et al. [19, 20]. A block diagram of the newly improved version of the spectrometer is shown in Fig. 1 and a detailed description of recent improvements is given by Helms et al. [21]. The microwave power was generated by reflex klystrons in the range from 28-40 GHz and multiplied by means of harmonic generation. The multiplier diodes used were phosphorus diffused p-type Si/W point contact diodes and were prepared and mounted in our laboratory using a slightly simplified variant of the method of Burrus [22, 17]. Used together with a recently installed liquid helium cooled InSb detector [21] they proved to generate sufficient power for measuring transitions up to 475 GHz. All measurements were carried out with the help of a digital data acquisition system [20] to improve the signal-to-noise ratio and to measure automatically the transition frequencies. The accuracy of the measurements is better than 20 kHz at a klystron fundamental frequency of 40 GHz. Sample pressure in the cell was in the range $20-60\,\mu bar$.

III. Assignment of the Measured Rotational Transitions

The available data on thioketene from the earlier measurements [7] enabled us to predict the entire ground-state rotational spectrum in the millimeter wave region with some certainty. The most intense absorption lines in the observed frequency range could easily be assigned to thioketene ground state transitions. The spectrum shows the characteristic pattern of a-type ^qR-branch rotational transitions. It differs only slightly from the ketene spectrum, having slightly less inertial asymmetry:

$$\varkappa(\text{CH}_2\text{CCS}) = -\,0.99918, \ \varkappa(\text{H}_2\text{CCO}) = -\,0.99722.$$

Figure 2 shows an oscilloscope display of the most intense lines of the $J = 10 \leftarrow 9$ transition together with the assignment of the ground state spectrum. The splitting of the $K_a = 1$ lines is 1.15 GHz, which places them outside the represented frequency range. Their intensity is comparable to the $K_a = 5$ line intensity. The intensities of the transitions follow basically an asymmetric top intensity distribution modulated by the spin statistics due to the two protons; because of the molecular C2V symmetry, transitions arising from odd Ka levels have three times the statistical weight of those arising from even K_a levels. Ground state transitions with $K_a > 7$ have intensities comparable to excited vibrational state transitions. Therefore the measurement and assignment of these weaker lines require precise predictions, which however could not be provided by the centrifugal distortion analysis (see discussion).

Table 1 gives the line frequencies of the observed and calculated transitions. It was impossible to resolve K_a doublets for $K_a > 2$. $K_a > 3$ and $K_a = 3$, J > 11 line frequencies were assigned and fitted to just one of the doublet transitions. For J > 10 the observed $K_a = 3$ lines were fitted to both components which were weighted equally. For J > 17 these lines

10wer

obs.-calc. / kHz

calculated Frequency / MHz

observed

TABLE I continued

168041,3202(42) 168001,336(35) 168001,3943(35) 167962,6424(34) 167911,5307(45)

. 7.2

179222.0540(46)

(67757.8196(56)

168041,3743 168001,3817 167001,3817 167910,5500^b 167910,5500^b 167757,7926 167757,7926 167757,7926 167757,7926 167757,7926 167757,7926 167757,7926 167757,7926

180152.9777(48) 179220.5366(56) 179244.3428(50)

79199.6699(37)

79199.6941 79199.6941

79199.7162(37) 79158.2144(36) 79103.6351(47) 79032.5936(60) 78939.6500(59) 90417.7021(56)

179158.1964 179103.6292 179032.6104

34.2 - 22.8 - 19.0 - 5.9

14.5

189456.9787(65) 191410.2768(54) 190418.8519(68)

190417.6901 189456.9969 191410.2698 190418.8616

90447.4166(59) 90397.6198(40) 90397.6963(40) 90353.4112(39) 90296.3529(50) 90219.8458(67) 01612.3094(67) 00597.9218(82) 02665.9704(66) 01616.6277(83) 701660.5444(70) 201595.2116(45)

90447.4020 90397.6463 90397.6463

Transition	Frequency / MMz	/ MHz	obscalc.		Tran
J (K K.) - J (K K.)	observed	calculated	/ kH2	/ ca-1	3 (K X
The second secon		(Standard Devietion)			
1(0.1)-0(0.0)		11203.9843(5)		0.000	. 9) 6
2 (0.2)-1(0.1)	22407.907	22407.9073(10)	6.0	0.374	9 (7 . 3
2(1,1)-1(1,0)	22522.222	22522.2303(15)	. 8.3	9.753	10 (0 , 1
2(1,2)-1(1,1)	22292.315	22292.3054(7)	9.6	9.750	10 (1.1
3(0,3)-2(0,2)	33611.7	33611.7076(16)	. 7.6	1.121	10 (1 , 9
1(1.3)-2(1.1)	33783 23	33438.3725(24)	2.5	10.493	10 (2 , 8
3(2.1)-2(2.0)	33607.84	33607.9116(11)	- 71.6	38.623	10 (3 . 8
3(2.2)-2(2.1)	33607.84	33607.7716(11)	68.5	38.623	10 (4 . 7
. (0.4)-3(0.3)		44815.3239(19)		2.242	10 (5 , 6
4(1.4)-3(1.3)		44584.3368(32)		11.609	10 (6.5
4(1,3)-3(1,2)		45044.1768(29)		11.632	
4 (2.3)-3(2.2)		44810.2138(14)		39.744	11 (0)
4(2.2)-3(2.1)		44810.5641(14)		39.744	
(3.5)-3(3.1)		44803.3621(15)		96.586	11 (2 . 1
5 (0.5) - 4 (0.4)		56018.6948(23)		3.737	11 (2.9
6 (1 . 5) - 4 (1 . 4)		55730.1642(38)		13.096	11 (3 . 9
5 (2.4)-4 (2.3)		56012.6291(17)		41.239	11 (3 . 8
\$ (2.3) - 4 (2.2)		56013.2297(17)		41.239	11 (• .
5 (3.3) - 4 (3.2)		56004.0415(18)		98.080	11 (5 , 7
\$ (4.2)-4(4.1)		55991.4536(18)		153.692	11 (6.6
(5 . 0) 5 - (9 . 0) 9	67221.7649	67221.7691(26)		\$.606	
.(1.6)-5(1.8)		66875.8205(44)		14.905	12 (0 , 1
(1.5)-6(1.4)	67565.5592	67565.5556(41)	3.6	15.012	12 (1 . 1
6(2.5)-5(2.4)	67214.6624	67214.6857(19)	-23.3	43.107	12 (2 , 1
) (67215.8787	67215.9117(20)	-33.0	43.107	12 (2 , 1
6 (3.4) - 5 (3.3)	67204.6026	67204.6130(21)	-10.4	19.94	12 (3 . 1
	67169 1808	67169.1792(27)	10.6	239.677	12 (3 . 9
					12 (4 . 8
7 (0 . 7) - 6 (0 . 6)		78424.4554(29)		7. 14	12 (5 . 8
7 (1.7) - 6 (1.6)		78625.9433(46)		17.266	12 (7 . 6
7 (2.6)-6 (2.5)		78416.6518(22)		45.349	13 (0)
7 (2.5) - 6 (2.4)		78418.6133(22)		45.349	13 (1 . 1
7 (3.5)-6 (3.4)	78405.0501	78405.0555(24)	• 5.4 •	92.190	13 (1 . 13
		78387.3788(24)		157.701	13 (2 , 13
7 (6.2) - 6 (6.1)		78332.6611(37)		344.453	13 (2 , 1)
. (0.0)-7(0.7)		89626.7225(31)		10.464	3 .
. (1.8)-7(1.7)		89166.4823(52)		19.788	13 (4 . 1
_		90086.0825(49)		19.895	13 (5,9
1 (2 . 6) - 7 (2 . 6)		89621.3378(24)		47.965	13 (7 . 7
.(3.6)-7(3.5)		89605.3474(26)		94.805	
. (4 . 5) - 7 (4 . 4)		89585.1071(26)		160.316	14 (1 , 1
		89568.0035(33)		244.431	14 (1 , 13
8 (6.2)-7 (6.1)		89522.5526(40)		347.066	14 (2 , 1
. 2) - 7 (89476.0934(53)			14 (2 , 15
	100828.5016	100828.4989(33)	2.7	13.454	14 (3 . 15
9(1.9)-8(1.8)	101345.9346	101345.9377(51)	. 3.1	22.900	
9(2.8)-8(2.7)	100819.8963	100819.8855(25)	10.8	\$0.954	14 (5 , 1
9 (2.7) - 8 (2.6)	100824.0773	100824.0884(26)	. 11.1	\$6.954	14 (6 . 9
9(3.7).8(3.6)	100805.4770	100805.4670(35)	10.0	97.79	14 (7 .
9 (4.6)-8 (4.5)	100782.6540	100762.6479(28)	: :	247.419	15 (0 . 1
					15 (1 . 1

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Transition	Frequ	Frequency / MHz	calc.	f. James	Transition
3 (K K.) - 3 (K K.)	observed	(Standard Deviation)	/ kHz	·• /	J (Kg; - Kg) - J (Kg · Kg)
	100212 2500	100212 2485/411		369 63	16 (2 , 13) - 14(2 , 12)
(7,3)-8(7,2	100659.9786	100659.9807(56)	2.1	471.102	15 (3 , 13) - 14(3 , 12)
10 0 101 - 0 0	112029 2254	112029 2276(34)	: :		16 (3 , 12) - 14(3 , 11)
	111456.0481	111456.0495(55)		26.108	- 14(4
10 (1.9)-9(1.8)	112605.4688	112605.4732(63)	:	26.281	15 (5 . 11) - 14(5 . 10)
10 (2,9)-9(2,8)	112021.0952	112021.0896(26)	9.6	54.317	16 (7,9)-14(7,8)
10 (3.8) - 9 (2.7)	112026.8519	112026.8684(27)	. 16.5	54.318	16 (0 , 16) - 15(0 , 15)
10 (4.7) - 9 (4.6)	111979.9963	111979.9777(29)	9.81	166.666	
10 (5,6)-9(5,5)	111946.0601	111946.0546(38)	5.5	250.779	16 (1 , 15) - 15(1 , 14)
10 (6.5)-9 (6.4)	111901.7216	111901.7245(45)	- 2.9	353.412	16 (2 , 16) - 15(2 , 14) 16 (2 , 14)
10 (7 . 4) - 9 (7 . 3)	111843.6209	111843.6476(57)	- 26.7	474.459	16 (3 , 14) - 16(3 , 13)
11 (0 , 11) - 10(0 , 10)	123230, 3256	123230.3351(34)	5.6	20.554	- 15(3 .
	123864.6482	123864.6535(53)	5.3	30.034	- 15(4 .
11 (2, 10) - 10(2,9)	123221.9685	123221.9762(28)	1.7	58.054	16 (6 . 10) - 15(6 . 9)
11 (2.9) - 10(2.8)	123229.6590	123229.6810(29)	- 22.0	58.054	16 (7,9)-15(7,8)
11 (3 . 9) - 10(3 . 8)	123206 1045	123205.1029(31)	9 0	104.893	17 (0 , 17) - 16(0 , 16)
11 (4.8) - 10(4.7)	123177.0763	123177.0731(31)	3.2	170.401	17 (1, 17) - 16(1, 16)
11 (5 . 7) - 10(5 . 6)	123139.7462	123139.7298(39)	16.4	254.513	
11 (6.6) - 10(6.5)	123090.9560	123090.9562(47)	- 0.2	357.144	17 (2 , 16) - 16(2 , 15) 17 (2 , 16) - 16(2 , 14)
. 10(7 .	123027.0579	123027.0697(58)	- 11.6	478.190	17 (3 , 16) - 16(3 , 14)
12 (0 , 12) - 11(0 , 11)	134430.2741	134430.2723(35)	9.1	24.664	17 (3 , 14) - 16(3 , 13)
(1.10-101.	135123.4397	135123.4429(52)	3.2	34.169	- 16(4 . 1
12 (2 , 11) - 11(2 , 10)	134422.5134	134422.6136(30)	- 0.2	62.164	17 (6 . 13) - 16(5 . 12)
12 (2 , 10) - 11(2 , 9)	134432.5323	134432.5293(31)	3.0	62.165	
12 (3 . 10) - 11(3 . 9)	134404.5764	134404.5758(32)	9.	109.003	- 17(0 .
12 (3 . 9) - 11(3 . 8)	134404.5764	134404.5890(32)	. 12.6	109.003	10 (1, 10) - 17(1, 17)
12 (5.8) - 13(5.7)	134333.1565	134333.1391(41)	17.4	268.621	. 1)(1.1
12 (6 . 7) - 11(6 . 6)	134279.9225	134279.9190(49)	3.5	361.250	10 (2 , 17) - 17(2 , 16)
12 (7,6) - 11(7,5)	134210.2101	134210.2224(57)	- 12.3	482.294	
(0 , 13) - 12(0 ,	145629.4681	145629.4739(35)	8.8	29.143	16 (3 . 15) - 17(3 . 14)
13 (1 , 13) - 12(1 , 12)	144887.7576	144887.7506(50)	7.0	38.377	- 17(4 . 13)
(2.12) - 12(2.	145622.6678	145622.6701(34)	5.3	58.676	. 17(5 .
- 12(2 .	145635.4592	145635.4165(33)	42.7	66.649	10 (7 , 12) - 17(7 , 11)
(11.1)-1	145603.8231	145603.7897(33)	33.4	113.486	19 (0 , 19) - 18(0 , 18)
13 (3 , 10) - 12(3 , 9)	145603.8231	145603.8095(33)	13.6	113.466	19 (1, 19) - 18(1, 18)
(6 . 8)	145526.2389	145526.2585(42)	. 19.6	263.102	19 (1 , 18) - 18(1 , 17)
- (8.4)	145468.5761	145468.5886(50)	- 12.5	365.728	19 (2, 13) - 18(2, 17)
13 (7,7)-12(7,6)	145393.0962	145393.0813(56)	14.9	486.771	19 (3 , 17) - 16(3 , 16)
14 (0 , 14) - 13(0 , 13)	156827.8757	156827.8788(37)	3.1	34.006	. 18(3 . 1
- 13(1 .	157639.7466	157639.7063(48)	40.3	43.559	19 (4 , 16) - 18(4 , 15)
. 13(2 .	156822.4152	156822.4139(39)	1.3	71.50\$	19 (6 . 14) - 18(6 . 13)
. 13(2 .	156838.3422	156838.3458(37)	3.6	71.507	19 (7, 13) - 16(7, 12)
14 (3 , 12) - 13(3 , 11)	156802.7331	156802.7229(34)	10.2	118.343	20 (0 . 20) - 19(0 . 19)
. 13(156766.7038	156766.7188(33)	. 15.0	183.848	:
14 (5 , 10) - 13(5 , 9)	156719.0705	156719.0637(43)	8.8	267.956	20 (1 , 19) - 19(1 , 18)
14 (6 . 9) - 13(6 . 8)	156656.9347	156656.9406(53)	6.3	370.582	
14 (7 , 8) - 13(7 , 7)	156575.6345	156575.6218(55)	12.7	491.621	20 (3, 18) - 19(3, 17) 2
15 (0 , 15) - 14(0 , 14)	168025.3840	168025.4258(40)	9.0	39.237	(3.17) - 19(3.16)
15 (1 . 14) - 14(1 . 13)	168897.1386	168897.1089(47)	29.7	4.8.7	20 (4 , 17) - 19(4 , 16) 2;
(2,14)-14(2,	168021.7032	168021.7133(47)	- 10.1	76.736	
-					20 (7, 14) - 19(7, 13)

10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02 | 10.02

9.6

90121.0886(67)

90121.1135 00597.9132

190253.4157 190296.3490 190219.8371

5.5

101406.6826(77)

201595,3198 201548,2060 201548,2060 201486,5863^b 201406,6726 201302,1208

(01302.1110(80) 112805.8149(82)

201595.3135(45) 201648.2098(43) 201486.6599(56)

10.2 0.7 72.0 61.8 11.9 25.0

213921.0328 21283.3335 21285.7392.4849 212792.4849 212792.4849 212792.573

211738.2618(105)

12813.8322(99) 12853.7292(84) 12792.4129(52) 112792.5467(52) 12742.5862(49) 12677.5319(63) 12593.0795(90) 12482.6926(98)

7.3

8.6

125176.3974(103) 24010.4336(119) 24056.9737(100)

(22877.9350(133)

(23998.1677(100)

12482.6931 23998.1650 26175.3888 . 91.6 . 91.6

23989.2018(61) 23989.3750(61) 23936.5173(57)

224010.7400 b 224056.9672 223969.2843 b 223956.5167 223956.5167

223867.9447(72) 223779.0121(106) 223662.8089(121)

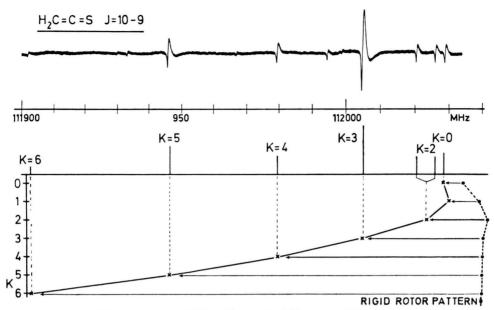


Fig. 2. Observed spectrum of thicketene at 112 GHz with a typical K_a pattern (K_a represented by K). The center of the $K_a = 1$ and 2 transitions is indicated in the diagram.

had to be rejected. Several other measured transitions were also rejected from the fit, indicating an error in measurement or overlap with excited state lines. The standard deviation of the least squares analysis was obtained to be 16.5 kHz, which corresponds to the assumed reproducibility of the frequency measurements.

IV. Centrifugal Distortion Analysis

The centrifugal distortion analysis of the observed rotational spectrum was performed using Watson's reduced Hamiltonian in the I^r axis representation [18]. As in previous papers on ketene [3, 4], the A-reduced form of the Hamiltonian operator with $R_6=0$ was used first to calculate the rotational energy levels. Further refinement of the distortion constants could not be achieved with this form. The higher inertial symmetry of thioketene made it necessary to use the S-reduced Hamiltonian according to Watson, in which R_5 is zero. The complete form of this operator with the constants used is given below:

$$\tilde{\mathfrak{G}} = \frac{1}{2} (B+C) \hat{P}^2 + [A - (B+C)/2] \hat{P}_z^2
- D_J \hat{P}^4 - D_{JK} \hat{P}^2 \hat{P}_z^2 - D_K \hat{P}_z^4 + H_{JK} \hat{P}^4 \hat{P}_z^2
+ H_{JK} \hat{P}^2 \hat{P}_z^4 - L_{KJ} \hat{P}^2 \hat{P}_z^6 + S_{KJ} \hat{P}^2 \hat{P}_z^8$$

$$egin{aligned} +\left[\,(B-C)/4+d_{1}
ight]\hat{P}^{2}\cdot[\,\hat{P}_{+}^{\,\,\,2}+\hat{P}_{-}^{\,\,2}]\ +d_{2}\,\,\,\hat{P}^{2}\cdot[\,\hat{P}_{+}^{\,\,4}+\hat{P}_{-}^{\,\,4}] \end{aligned}$$
 with $\hat{P}_{+}=\hat{P}_{x}\pm i\,P_{x}$.

 \hat{P} , \hat{P}_x , \hat{P}_y , and \hat{P}_z are the operators for the total angular momentum and its components. The constants A, B and C are Watson's reduced rotational constants in the I^r axis representation. Only two sextic constants have been used in our calculations. In addition to the sextic terms we found it necessary to add two higher order terms, L_{KJ} and S_{KJ} , in order to fit the observed transitions with K_a values of 6 and 7. The reduced spectroscopic constants in the I^r axis representation are given in Table 2.

As was pointed out by Nemes and Winnewisser [4], a-type rotational spectra of nearly symmetric prolate rotors contain information about the difference $(A-D_K)$, but not about A and D_K separately. In order to obtain the best possible A constant the value of D_K was taken from ketene [3] and was held fixed in the least squares analysis. The uncertainty of D_K might cause a larger error of A than that calculated from the least squares analysis and given in Table 2. The remaining rotational constants and the quartic and sextic constants are well determined. The octic constant L_{KJ} is poorly

Table 2. Spectroscopic parameters of thicketene for Watson's reduced Hamiltonian in the I^{r} -axis representation a.

A/MHz	286 655 (82)	
B/MHz	5 659.47596 (72)	
C/MHz	5 544.51269 (75)	
D_J/kHz	1.08569 (4)	
D_{JK}/kHz	168.269 (77)	
D_K/MHz	23.5 (fixed)	
$d_1/{ m Hz}$	-25.46 (68)	
$d_2/{ m Hz}$	— 5.21 (35)	
H_{JK}/Hz	0.716 (20)	
H_{KJ}/Hz	-408.7 (73)	
$L_{KJ}/{ m Hz}$	0.65(24)	
$S_{KJ}/{ m Hz}$	-0.0533(24)	

^a The standard deviation of the fit is 16.5 kHz for 125 equally weighted lines. Numbers in parentheses are standard errors.

determined and the dectic constant S_{KJ} shows that the Hamiltonian is not converging properly. These two constants therefore have to be considered only as fitting parameters to allow the measured frequencies of the $K_a = 6$ and $K_a = 7$ lines to be included in the calculations. In order to complete the distortion analysis, perpendicular rovibrational infrared transitions would have to be included.

V. Dipole Moment

The electric dipole moment for thioketene was determined from Stark effect measurements of the $2_{02} \leftarrow 1_{01}$ transition to be 1.01(3) D. This value places thioketene between the homologues ketene, with $\mu = 1.41$ D [3], and selenoketene with $\mu = 0.90$ D [9].

VI. Discussion

The reduced spectroscopic constants given in Table 2 can be transformed to the so-called determinable constants, $\mathfrak{A}, \mathfrak{B}, \mathfrak{S}, \mathfrak{T}_{aa}, T_{bb}, T_{cc}, T_1$ and T_2 , following the notation of Watson [18] and Yamada and Winnewisser [23]. These constants are invariant to a unitary transformation of the Hamiltonian and they are listed together with the derived inertial moments and the τ -defect ΔT_{cc} in Table 3. The errors of these parameters are estimated from the standard deviations obtained in the least squares analysis. The small value for the τ -defect ΔT_{cc} permits the application of any of the various

Table 3. Watson's determinable constants and moments of inertia for thioketene^a.

A/MHz	286 655 (82)	
$\mathfrak{B}/\mathrm{MHz}$	5 659.64633 (80)	
$\mathfrak{C}/\mathrm{MHz}$	5 544.68316 (83)	
$I_a/\mu { m \AA}^{2~{ m b}}$	1.76301 (50)	
$I_b/\mu { m \AA}^2$	89.294625 (12)	
$I_c/\mu { m \AA}^2$	91.146056 (13)	
Δ/μÅ ^{2 b}	0.08842 (53)	
T_{aa}/MHz	-23.66935 (8)	
T_{bb}/kHz	— 1.1470 (25)	
T_{cc}/kHz	— 1.0452 (25)	
T_1/kHz	-171.495 (79)	
$T_2*/\mathrm{kHz}^{\mathrm{e}}$	-4.2204 (34)	
$\varDelta T_{cc}/\mathrm{Hz}^{\mathrm{d}}$	2.3 (73) undefined	

^a Numbers in parentheses are estimated errors on the basis of the standard errors of Table 2.

planarity relations [23] in order to obtain the complete set of derived quartic distortion constants as was done in the case of ketene [4].

In the discussion of the previous section it was shown that the addition of higher order terms in the Hamiltonian was necessary to fit the $K_a=6$ and $K_a=7$ transitions. However, the inclusion of these terms did not prove useful in predicting the line positions of the $K_a=8$ lines, which could not be found at the expected positions. Though their intensities are rather low, we believe that the Hamiltonian used is insufficient to allow good frequency predictions outside the observed range of K_a states. No predictions for $K_a>7$ lines were therefore included in Table 1.

In a recent communication Yamada [24] has explained and analyzed the observed anomaly found in the ${}^{r}R_{5}$ and ${}^{r}R_{6}$ branches of the far infrared spectrum of HNCO. An interaction between the $K_{a}=6$ and 7 levels in the ground vibrational state and the $K_{a}=5$ and 6 levels in the lowest excited vibrational state is caused in HNCO by a centrifugal distortion correction term in the Hamiltonian. Although the K_{a} energy levels in the ground vibrational state of $H_{2}CCS$ are lower in energy $(A \sim 30 \text{ cm}^{-1} \text{ for HNCO}, 9.5 \text{ cm}^{-1} \text{ for H}_{2}CCS)$, the ratio of the rotational constant A to the energy of the lowest bending mode is probably similar for HNCO and $H_{2}CCS$. Thus the lowest bending mode of $H_{2}CCS$ can be expected to be involved in this type

b $\Delta = I_c - I_a - I_b$. c $T_2^* = T_2/(\mathfrak{A} + \mathfrak{B} + \mathfrak{C})$.

d $\Delta T_{cc} = T_{cc} - (T_2 - \mathfrak{C} T_1)/(\mathfrak{A} + \mathfrak{B}).$

of resonance for some value of K_a . The symmetry selection rules are analogous in the two cases. The network of Coriolis interactions in HNCO is similar to that in H_2CCO [5, 6] and therefore a corresponding set of interactions can be expected in H_2CCS , favoring an interaction with the ground state.

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